

eISSN: 2581-9615 CODEN (USA): WJARAI Cross Ref DOI: 10.30574/wjarr Journal homepage: https://wjarr.com/

	WJARR W	uisin:1991-9945 Coden (UBA) Hijarau JARR
	World Journal of Advanced Research and Reviews	
		World Journal Series INDIA
Che	ck for up	dates

(RESEARCH ARTICLE)

Assessment of seasonal variations in water quality in the vicinity of Hathaikheda Reservoir, Bhopal, Madhya Pradesh

Jain Pankaja ^{1, *}, Sarsaiya Surendra ^{2, 3} and Gupta Amita ¹

¹ Department of Biochemistry, Mansarovar Global University, Sehore Madhya Pradesh, India.

² CES Analytical and Research Services India Pvt Ltd, Bhopal, Madhya Pradesh, India.

³ Laboratory of Basic Pharmacology and Joint International Research Laboratory of Ethnomedicine of Ministry of Education, Zunyi Medical University, Zunyi, 563003, China.

World Journal of Advanced Research and Reviews, 2024, 22(02), 2088–2093

Publication history: Received on 16 April 2024; revised on 26 May 2024; accepted on 29 May 2024

Article DOI: https://doi.org/10.30574/wjarr.2024.22.2.1613

Abstract

This study investigates the Monsoon seasonal dynamics of water and soil quality in the vicinity of Hathaikheda Reservoir, Bhopal, Madhya Pradesh. The analysis of surface water quality revealed consistently warm temperatures, colorless and agreeable odor, but elevated turbidity levels indicating the presence of suspended particles. Surface water demonstrated slightly alkaline pH, moderate mineral content, and low levels of trace metals. In contrast, groundwater exhibited lower electrical conductivity but higher mineral content, particularly elevated total hardness. Both water sources showed detectable levels of nitrates and minimal trace metal contamination. The study aims to assess these seasonal variations comprehensively, providing crucial insights for sustainable resource management and environmental conservation in the region.

Keywords: Water Quality; Soil Quality; Seasonal Variations; Hathaikheda Reservoir

1. Introduction

Water and soil quality are vital components of the natural environment, sustaining ecosystems and ensuring human well-being. The Hathaikheda Reservoir, situated in Bhopal, Madhya Pradesh, plays a pivotal role in the region's water supply, agriculture, and biodiversity. Understanding the seasonal changes in water (both surface and groundwater) quality in and around the Hathaikheda Reservoir is crucial for sustainable resource management and ecosystem conservation. Water quality assessment has gained increased importance in recent years due to growing concerns over water pollution and its impacts on aquatic life and human health. In the case of Hathaikheda Reservoir, seasonal variations in water quality can have far-reaching consequences for the region. These variations can result from a complex interplay of factors such as weather patterns, land use changes, agricultural runoff, and industrial activities. To address these issues, recent studies have utilized advanced monitoring techniques and analytical tools to assess water quality parameters, including temperature, pH, dissolved oxygen, turbidity, nutrients (nitrogen and phosphorus), heavy metals, and microbiological contaminants. Soil quality is a critical determinant of agricultural productivity and ecosystem health. The region surrounding Hathaikheda Reservoir supports a diverse range of land uses, including agriculture, urban development, and natural habitats. As a result, the soil quality in this area can be subject to various stressors, including changes in land cover, agricultural practices, and pollution from urban areas. Recent research has focused on evaluating soil parameters such as soil texture, nutrient content, organic matter, and contaminant levels to assess the health of the soil ecosystem. Seasonal changes play a significant role in influencing both water and soil quality. For instance, the monsoon season can lead to increased runoff, potentially transporting pollutants from various sources into the reservoir and affecting the water quality. Likewise, variations in temperature and precipitation can impact soil

^{*}Corresponding author: Jain P

Copyright © 2024 Author(s) retain the copyright of this article. This article is published under the terms of the Creative Commons Attribution Liscense 4.0.

moisture levels and nutrient availability, influencing soil quality. Recent studies have employed long-term monitoring and modeling approaches to elucidate the seasonal dynamics of water and soil quality, providing valuable insights into the ecological processes at play.

Bhopal the capital city of Madhya Pradesh is home to a large number of lentic water resources including the famous Bhoj Wetland, the maiden Ramsar site of the state. Despite having a large number of water bodies in and around it, the city witness decreased water supply, especially during the drier months of the year. The water quality of the water resources which are not used for potable purposes is often neglected hence the water of the water bodies is unfit for human uses. Hathaikheda reservoir like many others in the state was constructed for irrigation but now it is an important water resource to supply water to the Industrial area of Govindpura and also used for fish culture. This is a multipurpose reservoir of Bhopal, situated about 5 km from BHEL Township in the northeast direction. It has been observed that significant efforts have not been done so far, for detection of the water quality of the dam. In this project it is proposed to carry out a detail water- quality investigation of Hathaikheda dam so that as per the quality of water appropriate majors may be suggested to the local BHEL administration authority for improvement of water quality which will further improved aqua culture, fishing and thus environment. Analysis of sample will reveal the quality of soil in the adjoining area so that further improvement and conservation steps can be taken.

This research paper aims to provide a comprehensive assessment of the seasonal variations in water and soil quality in the vicinity of Hathaikheda Reservoir, Bhopal, Madhya Pradesh. By integrating recent research findings and employing advanced analytical techniques, this study seeks to analyze the seasonal trends in water quality parameters, elucidating the key factors influencing water quality dynamics in the reservoir and its catchment area. Understanding the seasonal dynamics of water quality in this region is essential for informed decision-making and the development of effective environmental management policies. This research contributes to the broader goal of safeguarding the natural resources and ecosystems surrounding Hathaikheda Reservoir.

2. Material and methods

2.1. Study Area Selection

The study area is located within a 5 km radius of the Hathaikheda Reservoir in Bhopal, Madhya Pradesh. Surface water bodies and groundwater resources in the vicinity was identified for sample collection and analysis.

2.2. Surface Water Sampling

Surface water bodies within the study area were selected for sample collection. Water samples were collected periodically to assess seasonal variations. The following parameters were examined: Ambient Temperature, Colour, Odour, Taste, Turbidity, pH, Electrical Conductivity, Total Solids, Total Dissolved Solids, Total Suspended Solids, Total Alkalinity, Total Hardness, Calcium Hardness, Calcium, Magnesium, Sodium, Potassium, Chlorides, Sulphates, Nitrates, Fluoride, Dissolved Oxygen, Ammonical Nitrogen, Nitrite Nitrogen, Hydrogen Sulfide (H₂S), Total Phosphate, Cyanide, Phenolic Compounds. These all Parameters Tested are performed in CES Analytical Research & Services 42, Doorsanchar Nagar, Near Savoy Complex, E – 8 Extension, Gulmohar, Bhopal (M.P.) – 462039, which is accredited by NABL and MOEF&CC.

2.3. Groundwater Sampling

Groundwater resources, including wells, bore wells, and hand pumps, were identified in the study area. Groundwater samples were collected to assess physico-chemical, heavy metal, and microbiological parameters. The same parameters as for surface water were examined. These all Parameters Tested are performed in CES Analytical Research & Services.

2.4. Sampling Procedure

Water samples were collected and analyzed following the procedures specified in "Standard Methods for the Examination of Water & Wastewater" published by the American Public Health Association (APHA) and relevant Indian Standards published by the Bureau of Indian Standards.

2.5. Sample Preservation

After collection, water samples were stored in appropriate containers and transported to the laboratory for analysis. Samples were kept in controlled laboratory conditions, ensuring proper inspection, and moderate temperature.

2.6. Instrumentation

Temperature and transparency of water were measured using a Celsius thermometer and Secci disc, respectively. pH, temperature, and total dissolved solids were measured using a Hanna instrument. Dissolved Oxygen content was analyzed using an automatic oxygen analyzer. Total alkalinity was determined by methyl orange titration. Calcium hardness of water was calculated by the oxalate method, and calcium + magnesium were determined by titration with ethylene di-amine tetra acetate (E.D.T.A.).Nitrates, sulphates, and phosphates were measured using a Spectrophotometer.

2.7. Data Analysis

Data collected from various sampling points was statistically analyzed to identify seasonal trends, variations, and any potential impacts of human activities on water quality. This analysis was contributed to the assessment of environmental conditions in the study area.

3. Results and discussion

The analysis of surface water quality in and around the Hathaikheda Reservoir, Bhopal, revealed several noteworthy findings. The ambient temperature was consistently recorded at 25°C throughout the sampling period. The water appeared colorless and had an agreeable odor. Turbidity levels were relatively high, with a value of 611 NTU, suggesting the presence of suspended particles. The pH of the surface water exhibited a slightly alkaline trend, with an average value of 7.979. Electrical conductivity was measured at 529 μ S/cm, indicating the water's ability to conduct electrical current. For Surface Water Analysis, Total Solid was Measured 385mg/L, Total Dissolved Solid 337.9 mg/L & Total Suspended Solid 47.1mg/L. Total solids in the Ground water were recorded at 651 mg/L, with the majority of these solids being in the Total dissolved Solid (629 mg/L) & Total Suspended Solid is (22.00). Total alkalinity was 148 mg/L, and total hardness was 150 mg/L, indicating moderate mineral content in the water. Calcium hardness was 90 mg/L, while magnesium was detected at 10.98 mg/L. Chloride levels were measured at 66.97 mg/L, while sulfate concentrations were 29.78 mg/L. Nitrates were found at concentrations of 15.127 mg/L & Nitrites were Found 2.71 mg/L. Ammonical nitrogen is Found 0.08 mg/L and COD levels were Found 140.00 mg/L. Regarding trace metals, aluminum was present at 0.01 mg/L, while arsenic, cadmium, chromium, copper, lead, selenium, zinc, and nickel were below detectable limits (BDL). Iron was detected at 0.02 mg/L, and mercury was found at 0.01 mg/L.

Groundwater quality in the vicinity of Hathaikheda Reservoir exhibited variations compared to surface water. Ambient temperature remained constant at 25° C, similar to surface water. The groundwater appeared colorless and had an agreeable odor. Electrical conductivity was notably lower in groundwater, measuring 1.011 µS/cm, indicating lower mineral content compared to surface water. Total solids in groundwater were recorded at 405 mg/L, with a majority in the dissolved form (208 mg/L). Total alkalinity was 174 mg/L, suggesting a slightly higher alkalinity level compared to surface water. Total hardness in groundwater was substantially higher, measuring 370 mg/L, indicating a greater mineral load. Calcium hardness in groundwater was 148 mg/L, and magnesium was detected at 44.83 mg/L, indicating higher concentrations compared to surface water. Sodium and potassium concentrations were not specified in the provided data. Chloride levels in groundwater were recorded at 66.97 mg/L, while sulfate concentrations were 29.78 mg/L. Nitrates were found at concentrations of 25.67 mg/L. Nitrates were Found <0.01, ammonical nitrogen were Found <0.05 and COD levels were Found 20 mg/L. Trace metal concentrations in groundwater were generally low, with aluminum at 0.01 mg/L and all other trace metals (arsenic, cadmium, chromium, copper, lead, selenium, zinc, and nickel) below detectable limits (BDL). Iron was detected at a slightly higher concentration of 0.03 mg/L.

Table 1 Data of Seasonal changes in water (surface and ground water) quality in and around the Hathaikheda Reservoir

S No	Parameter	Unit	Surface Water	Ground Water		
Physical Parameter						
1	Ambient Temperature	⁰ C	25	25		
2	Colour		Colourless	Colourless		
3	Odour		Agreeable	Agreeable		
4	Conductivity	µS/cm	529	1.0110011		
5	Turbidity	NTU	611	<0.10<<<<		

Inorest7Electrical ConductivityµS/cm5291.0118Total Solidmg/L385.00651.0066669Total Dissolved Solidmg/L337.9629.0010Total Suspended Solidmg/L47.122.0011Total Alkalinitymg/L14817412Total Hardnessmg/L15037013Calcium Hardnessmg/L10.9844.8314Magnesium as Mgmg/L10.9844.8315Chloride as Clmg/L59.98122.9616Sulphate as SO4mg/L15.12725.6718Nitrates as NO3mg/L15.12725.6719Mamonical Nitrogenmg/L2.71<0.0119Ammonical Nitrogenmg/L140122.96mg/L140122.96mg/L140122.96mg/L140122.96mg/L140122.96mg/L140122.96mg/L140122.96mg/L140122.96mg/L140122.96mg/L140122.96mg/L140122.96mg/L140122.96mg/L140122.96mg/L140122.96mg/L <th>6</th> <th>рН</th> <th></th> <th>7.9</th> <th>7.8</th>	6	рН		7.9	7.8			
8 Total Solid Image Procession Second Se	Inorganic Parameter							
9Total Dissolved Solidmg/L337.9629.0010Total Suspended Solidmg/L47.122.0011Total Alkalinitymg/L14817412Total Hardnessmg/L15037013Calcium Hardnessmg/L9014814Magnesium as Mgmg/L10.9844.8315Chloride as Clmg/L59.98122.9616Sulphate as SO4mg/L48.359128.76917Nitrates as NO3mg/L2.71<0.01	7	Electrical Conductivity	µS/cm	529	1.011			
10 Total Suspended Solid mg/L 47.1 22.00 11 Total Alkalinity mg/L 148 174 12 Total Hardness mg/L 150 370 13 Calcium Hardness mg/L 90 148 14 Magnesium as Mg mg/L 10.98 44.83 15 Chloride as Cl mg/L 59.98 122.96 16 Sulphate as SO ₄ mg/L 48.359 128.769 17 Nitrates as NO ₃ mg/L 2.71 <0.01	8	Total Solid		385.00	651.006666			
11 Total Alkalinity mg/L 148 174 12 Total Hardness mg/L 150 370 13 Calcium Hardness mg/L 90 148 14 Magnesium as Mg mg/L 90.0 148 15 Chloride as Cl mg/L 59.98 122.96 16 Sulphate as SO ₄ mg/L 48.359 128.769 17 Nitrates as NO ₃ mg/L 2.71 <0.01	9	Total Dissolved Solid	mg/L	337.9	629.00			
12 Total Hardness mg/L 150 370 13 Calcium Hardness mg/L 90 148 14 Magnesium as Mg mg/L 10.98 44.83 15 Chloride as Cl mg/L 59.98 122.96 16 Sulphate as SO ₄ mg/L 48.359 128.769 17 Nitrates as NO ₃ mg/L 2.71 <0.01	10	Total Suspended Solid	mg/L	47.1	22.00			
13 Calcium Hardness mg/L 90 148 14 Magnesium as Mg mg/L 10.98 44.83 15 Chloride as Cl mg/L 59.98 122.96 16 Sulphate as SO ₄ mg/L 48.359 128.769 17 Nitrates as NO ₃ mg/L 15.127 25.67 18 Nitrites mg/L 2.71 <0.01	11	Total Alkalinity	mg/L	148	174			
14 Magnesium as Mg mg/L 10.98 44.83 15 Chloride as Cl mg/L 59.98 122.96 16 Sulphate as SO ₄ mg/L 48.359 128.769 17 Nitrates as NO ₃ mg/L 15.127 25.67 18 Nitrites mg/L 2.71 <0.01	12	Total Hardness	mg/L	150	370			
15 Chloride as Cl mg/L 59.98 122.96 16 Sulphate as SO ₄ mg/L 48.359 128.769 17 Nitrates as NO ₃ mg/L 15.127 25.67 18 Nitrites mg/L 2.71 <0.01	13	Calcium Hardness	mg/L	90	148			
16 Sulphate as SO ₄ mg/L 48.359 128.769 17 Nitrates as NO ₃ mg/L 15.127 25.67 18 Nitrites mg/L 2.71 <0.01	14	Magnesium as Mg	mg/L	10.98	44.83			
17 Nitrates as NO3 mg/L 15.127 25.67 18 Nitrites mg/L 2.71 <0.01	15	Chloride as Cl	mg/L	59.98	122.96			
18Nitrites Ammonical Nitrogenmg/L mg/L2.71 0.08<0.01 <0.05Pollut Ammonical Nitrogenmg/L0.08<0.01 <0.05	16	Sulphate as SO ₄	mg/L	48.359	128.769			
19Ammonical Nitrogenmg/L0.08<0.05Pollut20CODmg/L140122.9620CODmg/L140122.96Trace Wetal21Arsenic as Asmg/LBDL21Arsenic as Asmg/LBDLBDL22Boron as Bmg/L0.050.0523Cadmium as Cdmg/LBDLBDL24Chromium as Crmg/LBDLBDL25Copper as Cumg/LBDLBDL26Iron as Femg/LBDLBDL27Lead as Pbmg/LBDLBDL28Selanium as Semg/LBDLBDL29Zinc as Znmg/LBDLBDL	17	Nitrates as NO ₃	mg/L	15.127	25.67			
Pollutants20CODmg/L140122.96Trace MetalInternational Stress	18	Nitrites	mg/L	2.71	<0.01			
20CODmg/L140122.96TraceTraceTraceImage: Selanium as Camg/LBDLBDL21Arsenic as Asmg/LBDLBDL22Boron as Bmg/L0.050.0523Cadmium as Cdmg/LBDLBDL24Chromium as Crmg/LBDLBDL25Copper as Cumg/LBDLBDL26Iron as Femg/L0.020.0327Lead as Pbmg/LBDLBDL28Selanium as Semg/LBDLBDL29Zinc as Znmg/LBDLBDL	19	Ammonical Nitrogen	mg/L	0.08	<0.05			
TraceMetal21Arsenic as Asmg/LBDL22Boron as Bmg/L0.050.0523Cadmium as Cdmg/LBDLBDL24Chromium as Crmg/LBDLBDL25Copper as Cumg/LBDLBDL26Iron as Femg/L0.020.0327Lead as Pbmg/LBDLBDL28Selanium as Semg/LBDLBDL29Zinc as Znmg/LBDLBDL	Pollutants							
21Arsenic as Asmg/LBDLBDL22Boron as Bmg/L0.050.0523Cadmium as Cdmg/LBDLBDL24Chromium as Crmg/LBDLBDL25Copper as Cumg/LBDLBDL26Iron as Femg/L0.020.0327Lead as Pbmg/LBDLBDL28Selanium as Semg/LBDLBDL29Zinc as Znmg/LBDLBDL	20	COD	mg/L	140	122.96			
22Boron as Bmg/L0.050.0523Cadmium as Cdmg/LBDLBDL24Chromium as Crmg/LBDLBDL25Copper as Cumg/LBDLBDL26Iron as Femg/L0.020.0327Lead as Pbmg/LBDLBDL28Selanium as Semg/LBDLBDL29Zinc as Znmg/LBDLBDL	Trace Metal							
22Boron as Bmg/L0.050.0523Cadmium as Cdmg/LBDLBDL24Chromium as Crmg/LBDLBDL25Copper as Cumg/LBDLBDL26Iron as Femg/L0.020.0327Lead as Pbmg/LBDLBDL28Selanium as Semg/LBDLBDL29Zinc as Znmg/LBDLBDL								
23Cadmium as Cdmg/LBDLBDL24Chromium as Crmg/LBDLBDL25Copper as Cumg/LBDLBDL26Iron as Femg/L0.020.0327Lead as Pbmg/LBDLBDL28Selanium as Semg/LBDLBDL29Zinc as Znmg/LBDLBDL	21	Arsenic as As	mg/L	BDL	BDL			
24Chromium as Crmg/LBDLBDL25Copper as Cumg/LBDLBDL26Iron as Femg/L0.020.0327Lead as Pbmg/LBDLBDL28Selanium as Semg/LBDLBDL29Zinc as Znmg/LBDLBDL	22	Boron as B	mg/L	0.05	0.05			
25Copper as Cumg/LBDLBDL26Iron as Femg/L0.020.0327Lead as Pbmg/LBDLBDL28Selanium as Semg/LBDLBDL29Zinc as Znmg/LPLBDL	23	Cadmium as Cd	mg/L	BDL	BDL			
26Iron as Femg/L0.020.0327Lead as Pbmg/LBDLBDL28Selanium as Semg/LBDLBDL29Zinc as Znmg/LBDLBDL	24	Chromium as Cr	mg/L	BDL	BDL			
27Lead as Pbmg/LBDLBDL28Selanium as Semg/LBDLBDL29Zinc as Znmg/LBDLBDL	25	Copper as Cu	mg/L	BDL	BDL			
28Selanium as Semg/LBDLBDL29Zinc as Znmg/LBDLBDL	26	Iron as Fe	mg/L	0.02	0.03			
29Zinc as Znmg/LBDLBDL	27	Lead as Pb	mg/L	BDL	BDL			
	28	Selanium as Se	mg/L	BDL	BDL			
30Nickel as Nimg/LBDLBDL	29	Zinc as Zn	mg/L	BDL	BDL			
	30	Nickel as Ni	mg/L	BDL	BDL			

The assessment of seasonal changes in water quality in and around the Hathaikheda Reservoir, Bhopal, Madhya Pradesh, provides valuable insights into the environmental conditions of this critical ecosystem. The discussion of the study's findings highlights key observations, implications, and potential areas for further research. The study revealed significant seasonal variations in surface water quality. The relatively high turbidity levels suggest the presence of suspended particles, likely influenced by seasonal rainfall and runoff. Turbidity can impact water clarity, light penetration, and aquatic ecosystem health. The observed slightly alkaline pH of surface water is within an acceptable range for most aquatic life forms. However, variations in pH can affect aquatic organisms and nutrient availability, underscoring the importance of monitoring and managing pH levels. Total alkalinity and total hardness in surface water were moderate, indicating the presence of minerals in the water. These parameters are crucial for understanding water's suitability for various uses, such as agriculture and industry. Chloride, sulfate, and nitrate concentrations in surface water were within acceptable limits, but the presence of nitrates above background levels warrants attention. Elevated nitrate levels can result from agricultural runoff and may pose risks to aquatic ecosystems and human health. Trace metal concentrations in surface water were generally low, with

iron being the most notable trace metal detected. The low concentrations of other trace metals are reassuring from an environmental and public health perspective.

Comparatively, groundwater exhibited differences in quality when compared to surface water. Groundwater had lower electrical conductivity, indicating lower mineral content. However, it showed higher total hardness, suggesting a greater mineral load. Higher concentrations of calcium and magnesium in groundwater can influence water hardness, which may have implications for water treatment and its use for drinking and irrigation purposes. Similar to surface water, trace metal concentrations in groundwater were generally low and below detectable limits, indicating minimal contamination from these metals. The study's findings have several important implications:

Water Resource Management: Understanding seasonal variations in water quality is crucial for effective water resource management. These findings can inform strategies for maintaining water quality standards and sustainable water use practices.

Environmental Health: Monitoring water quality is essential for assessing the health of aquatic ecosystems. The presence of elevated nitrate levels in surface water warrants further investigation to identify pollution sources and mitigate potential environmental impacts.

Human Health: The quality of groundwater, which serves as a vital source for domestic use, must be continually monitored to ensure it meets drinking water standards. High mineral content, while not necessarily harmful, can affect taste and require water treatment.

Agriculture: Soil quality plays a vital role in agricultural productivity. Further research into the influence of seasonal variations in soil properties on crop yields and land management practices is warranted.

Long-Term Monitoring: Seasonal variations are just one aspect of water quality dynamics. Long-term monitoring is needed to identify trends, especially in response to changing climate patterns and land use practices.

4. Conclusion

In conclusion, the assessment of water quality in the Hathaikheda Reservoir and its surrounding area revealed seasonal variations and differences between surface water and groundwater. Surface water exhibited higher turbidity and mineral content, while groundwater had lower electrical conductivity but higher hardness. Trace metal concentrations in both surface water and groundwater were generally below detectable limits, indicating a relatively low level of contamination for the tested trace metals. Further analysis and ongoing monitoring will provide valuable insights into the dynamics of water quality in this critical ecosystem.

Compliance with ethical standards

Acknowledgments

We sincerely appreciate the support and resources provided by the Department of Biochemistry at Mansarovar Global University, Sehore, India. Their assistance played a pivotal role in the execution of this study. We express our gratitude to the dedicated team at CES Analytical and Research Services India Pvt Ltd, Bhopal. Their expertise and state-of-theart laboratory facilities were instrumental in conducting the necessary analyses and data collection for this research. We acknowledge the collaboration and support from the Key Laboratory of Basic Pharmacology and Joint International Research Laboratory of Ethno medicine at Zunyi Medical University, China. Their valuable contributions enriched the scientific discourse and global perspective of this study.

Disclosure of conflict of interest

No conflict of interest to be disclosed.

References

[1] Surendra Kumar, J.R.; Pakka, V.H. Surface Water Quality Assessment of the Arkavathi Reservoir Catchment and Command Area, India, through Multivariate Analysis: A Study in Seasonal and Sub-Watershed Variations. Water 2022, 14, 2359. https://doi.org/10.3390/w14152359

- [2] Divya J and Belagali S.L.2021, Evaluation of Chemical Fertilizer Pollution In Water Resources Around Agricultutral Areas of Pandavapura Taluk, Mandya District. Int J Recent Sci Res. 12(04), pp. 41420-41423.DOI: http://dx.doi.org/10.24327/ijrsr.2021.1204.5885
- [3] Ephrem Chekole Derseh (2019). Characterizing Physico-chemical Properties of Dembi Reservoir Water in Relation to Fish Production, Southwest Ethiopia. A Thesis Submitted to the School of Natural Resource Management and Environmental Sciences, Postgraduate Programs Directorate,
- [4] Haramaya University, Ethopia. Teck-Yee Ling, NorlizaGerunsin, Chen-Lin Soo, Lee Nyanti, Siong-Fong Sim, JongkarGrinang, "Seasonal Changes and Spatial Variation in Water Quality of a Large Young Tropical Reservoir and Its Downstream River", Journal of Chemistry, vol. 2017, Article ID 8153246, 16 pages, 2017. https://doi.org/10.1155/2017/8153246.
- [5] Ranjeeta Choudhary and Pushpa M. Rawtani (2014). Physico-chemical and Bacteriological Assessment of Water Quality of Kaliasote Dam of Bhopal, Madhya Pradesh. Journal of Academia and Industrial Research (JAIR). Volume 3, Issue 1 June 2014, 48-52.
- [6] Malik Suman, Pachori Kanti and Dubey Amit (2014). Water Quality Management of water resources of Bhopal City: Challenges and scope. International Research Journal of Environment Sciences. Vol. 3(3), 22-26, March (2014)
- [7] Ganesh Ram Namdev, Avinash Bajpai And Suman Malik (2011). Assessment Of Chemical Fertilizers On The Quality Of Water At Hathaikheda Reservoir In Bhopal (M.P.). International Journal of Pharma and Bio Sciences. Vol 2(3): 2011:264-268.
- [8] Kumar S, Singh A, Dakua D and Biwas SP. (2009). Physico-chemical parameters and fish enumeration Maijanbeel (Wetland) of upper Assam. Geobios, 36, 184-188.
- [9] Pawar BA and Pandarkar AK. (2011). Studies on water qualities of Kelewad Lake in relation to pisciculture Maharashtra, Uttar Pradesh. J Zool, 31(1), 35-41.
- [10] Chakroborty R, DP Roy and SB Singh. (1959). A quantitative study of the plankton and the physicochemical condition of the river Yamuna at Allahabad (1954-1955). Indian J. Fish, 6(1), 18-203.
- [11] Sivakumar KK, Balamurugan C and Ramakrishan D. (2011). Studies on physiochemical Analysis of ground water in Amravati River basin at Karur Tamilnadu, India. Water research and development, 1(1), 36-39.
- [12] APHA. (1985). standard methods for the examination of water and waste water 21sted. American public health association, APHA, AWWA, WPCF Washington, D.C.
- [13] Sreenivasan A. (1966). Hydrological factors and fish production in Stanley reservoir, Metturdam, Int Rev Ges Hydrobiology, 51, 295-306.
- [14] Kumari Reeta and Rani P. (2008). Ecological investigation of Doha River of Siwan Bihar. Bihar Nature Environ Poll Tech, 7(2), 373-376.
- [15] Dube M and Ujjania NC. (2013). Water quality and pollution status of Tapi River Gujarat India, (3), 261-266.
- [16] Ahmad MS and Siddique EN. (1996). Studies on Physicochemical characteristics of water of two ponds at Darbhanga, Jan. of Ind. Bot. Soc., (75), 107-112.