

eISSN: 2581-9615 CODEN (USA): WJARAI Cross Ref DOI: 10.30574/wjarr Journal homepage: https://wjarr.com/

WJARR	NISSN 2591-9915 CODEN (USA): WJARAJ
W	JARR
World Journal o	e L
Research and Reviews	6
	World Journal Series INDIA

(RESEARCH ARTICLE)

Check for updates

Mineralogical, geochemical and physical properties assessment of clay deposits in Umuoke Obowo Southeastern Nigeria for industrial applications

Thaddeus. C. Azubuike ^{1,*}, Paulinus. N. Nnabo ², Norbert Okechinyere Osonwa ¹, Chukwuemeka Emmanuel Odoala ³, Emma Onochie Nwabineli ¹ and Victor Dorawa Koreyo ¹

¹ Department of Ceramic and Glass Technology, Akanu Ibiam Federal Polytechnic, Unwana, Ebonyi State, Nigeria.
 ² Department of Geology, Ebonyi State University, Abakaliki Ebonyi State, Nigeria.
 ³ Department of Metallurgical Engineering Technology, AkanuIbiam Federal Polytechnic, Unwana, Ebonyi State, Nigeria.

World Journal of Advanced Research and Reviews, 2024, 21(03), 533-545

Publication history: Received on 22 January 2024; revised on 02 March 2024; accepted on 04 March 2024

Article DOI: https://doi.org/10.30574/wjarr.2024.21.3.0699

Abstract

This study assessed the suitability of clay deposits from Umuoke, Obowo in southeastern Nigeria as local raw materials for industrial applications. Ten samples were collected from different mining pits across the locality. X-ray fluorescence, X-ray diffraction and physical property tests characterized the clays based on geochemical composition, mineralogy and key attributes. The clays showed high silica (56.41%) and alumina (32.82%) contents typical of aluminosilicate clays. Iron oxide levels were moderately elevated (3.34% Fe₂O₃). Clay minerals kaolinite (18.9-37.0%) and illite (0.5-4.15%) occurred predominantly alongside non-clay minerals like quartz, feldspars and metal oxides. The clays exhibited high plasticity (avg. plasticity index 26.61%) enabling easy moulding and shaping. Porosity averaged 21.33% appropriate for refractories. Firing shrinkage (6.5-19.2%) and density (1.54-1.76 g/cm³) were in acceptable ranges. Strength postfiring reached the 15 N/mm² minimum standard. Estimated refractoriness was 1680.22°C. Overall, the Umuoke clays demonstrate favourable chemistry, mineralogy and physical properties for refractories and structural ceramics applications pending some processing adjustments. Locally exploiting these deposits can promote import substitution, rural industrialization and sustainable development in Nigeria. Further pilot testing can optimize formulations and processes for targeted ceramic products. Comprehensive nationwide clay deposit prospecting is also recommended. The clays are suitable for refractory bricks, ceramic tableware, architectural ceramics, wall tiles, pottery items.

Keywords: Geochemical composition; Mineralogy; X-ray fluorescence; X-ray diffraction.

1. Introduction

Nigeria possesses vast, yet largely untapped, clay deposits with potential economic benefits (1). These industrial clay resources could significantly boost Nigeria's economy and reduce import dependency if properly assessed and utilized by local industries (2). Despite the government's historical focus on exploiting these resources, issues surrounding the perceived unsuitability of available deposits have impeded progress (3,4). To address this, further survey and characterization of Nigerian clay deposits are urgently needed to determine their properties and industrial applicability. The hindrance to beneficiation and application of Nigerian clays arises from misconceptions about their mineralogical, chemical, and physical characteristics (3,4). Despite past research demonstrating successful uses of local clays, exploitation remains below potential due to persistent perception issues. Consequently, a comprehensive survey and characterization of the diverse clay deposits across Nigeria are crucial for accurate assessment of their properties and industrial suitability.

Copyright © 2024 Author(s) retain the copyright of this article. This article is published under the terms of the Creative Commons Attribution Liscense 4.0.

^{*}Corresponding author: Thaddeus. C. Azubuike , Email: azubuikethaddy@yahoo.com

Proper evaluation of mineralogical, chemical, and physical attributes is essential as clay utilization feasibility depends on these qualities for intended applications (5,6). Clay minerals play a significant role in determining the properties and industrial value of deposits (7). Kaolinitic clays are valuable for ceramics manufacturing and various industries due to their desirable properties (8). On the other hand, smectite and illite clays find applications in pharmaceuticals, cosmetics, oil drilling, wastewater treatment, and industrial processes (9). Assessing Nigerian clay typology and attributes is critical for scientific and socioeconomic perspectives to understand their composition-property-function relationships and potential economic contributions.

Imo state in Nigeria's southeast region harbors underexplored clay deposits in Umuoke Obowo Local Government Area with potential industrial uses (10). Despite a lack of prior documentation on the mineralogy, chemistry, and physical properties of these deposits, the study aims to fill this knowledge gap. The potential industrial applications of Umuoke Obowo clays, if found suitable, could contribute to import substitution, rural industrialization, and employment. The study proposes a methodology using modern analytical techniques like X-ray diffraction (XRD) and X-ray fluorescence (XRF) for evaluating obscure Nigerian clay deposits against technical specifications. If Umuoke Obowo deposits, including kaolinite and ball clays, are substantiated as suitable sources through rigorous characterization, they could reduce Nigeria's dependence on foreign kaolin and ceramic imports, providing a model approach for mapping and utilizing the country's clay reserves during economic uncertainty.

1.1. Geological Setting

The study area contains sediments of the Benin and Ogwashi-Asaba formations, southeast Nigeria (11). The Benin formation outcrops in Avutu Obowo, comprising cross-bedded sands and clays, up to 2100m thick, dipping southwest. The underlying Ogwashi-Asaba formation spans the Oligocene-Miocene with blue clays, shales, silts and sands containing lignite seams. It dips northeast with 450-900m thickness (12).

Ten clay samples were collected 4-10m deep in Umuoke after removing overburden (11). The Umuoke area specifically has clay-sandstone interbeds, white to brown in color. Sampling sites spanned 5°33'30.61''N to 5°34'53.02''N and 7°22'32.87''E to 7°23'24.80''E based on GPS. The lithology matches previous Oligocene-Miocene descriptions, likely representing Anambra basin sedimentation.

2. Materials and methods

2.1. Materials

The materials utilized during fieldwork and sampling included topographic maps, a Brunton compass, clinometer, hammer, spade, measuring tape, sample bags, digital camera and GPS device. For sample preparation, a jaw crusher, sieve, grinding equipment and sealing bags were used. The laboratory equipment for chemical and mineralogical characterization included an X-Ray Fluorescence Spectrometer (XRF), X-Ray Diffractometer (XRD), sample mill, pressing equipment and oven. Materials needed for the physical testing comprised plasticity cans, grooving tools, kiln, electronic balance, stove and a hydraulic press.

2.2. Study Area

The study area encompasses clay exposures in existing sand mining pits across the Umuoke locality, Obowo LGA in Imo State, southeastern Nigeria. Ten sampling points were selected from latitudes 5°33'30.61''N to 5°34'53.02''N and longitudes 7°22'32.87''E to 7°23'24.80''E. Fig. 4 shows the distribution of sampling sites within Umuoke.

2.3. Field Sampling

Clay samples were collected from depths of 4-10m across designated mining pits after removing the overburden layer. The reddish-brown overburden comprises lateritic soil, hardened sandstone and/or loose sand. At each pit, the clay unit appeared as white, light grey or brown massive bedrock, occasionally alternating with friable sandstones. The fine-grained, smooth clay sediments were extracted using picks and hammers based on accessibility.

2.4. Sample Preparation

The clay samples were air-dried for two weeks, then mechanically crushed using a jaw crusher and ground into fine powder below 150 microns grain size. About 40 grams of the powdered subset was stored in sealed bags and transported to the laboratory for chemical and mineralogical analyses. The remaining bulk sample was retained for physical testing. Prior to each new sample, the crusher was thoroughly cleaned to avoid cross-contamination.

2.5. Chemical Analysis

The major oxide composition of the clay fraction was quantified using X-Ray Fluorescence (XRF) Spectrometry. XRF tested major oxide composition. Samples milled below 150 microns. 30-40g powder in sample cups. XRF operated 35-40kV. Diffraction spectra collected. Peaks matched to database oxide wt % output.

2.6. Mineralogical Analysis

The bulk mineralogy of powdered clay specimens was determined using XRD at the National Steel and Raw Materials Exploration Agency. XRD determined mineralogy. Samples milled below 0.15 microns. XRD parameters: 2-60° 2theta, 6°/min, 40V, 30A. Smearing on holders, automated analysis and background noise removal Matching were attached to database to accept mineral fractions output.



Figure 1 Sample Location Map of the Studied Umuoke Obowo Area

2.7. Physical Testing

A range of tests were undertaken to evaluate the practical performance characteristics and industrial applicability of the Umuoke clay deposits as summarized below. Tests evaluated practical performances were liquid limit, plastic limits and plasticity index per ASTM D4318. Firing shrinkage of molded bars was determined per ASTM C326. Apparent porosity, water absorption, bulk density, and apparent specific gravity were evaluated per ASTM C20. Compressive strength of cubic specimens were determined while the refractoriness prediction was determined using chemical oxide content.

Results compiled, averaged, compared to standards to assess suitability and favorable applications.

3. Results

3.1. Chemical Analysis

The chemical analysis of the Umuoke clay samples was conducted using X-ray fluorescence (XRF) spectroscopy to determine the oxide compositions. The Umuoke clays were found to be predominantly composed of silica (SiO₂) and

alumina (Al_2O_3), with average concentrations of 56.41% and 32.82%, respectively (Table 1). These elevated SiO₂ and Al_2O_3 levels classify the Umuoke deposits as hydrated alumino-silicates (13,14).

Additionally, notable levels of iron oxide (Fe₁O₃) averaging 3.46% were detected across the Umuoke samples. The presence of Fe₂O₃ is likely attributable to the effects of superficial oxidation and contamination by Fe-rich solutions percolating from the overlying ferruginous sandstone and laterite. Meanwhile, concentrations of other major oxides, including TiO₂, CaO, MgO, Na₂O and K₂O were relatively minor, altogether constituting less than 5% of the average composition. The loss on ignition (LOI) values was consistently below 3%, indicating minimal organic matter in the clays.

SP										
Code	0k1	OK 2	OK 3	0K 4	OK 5	0K 6	0K 7	OK 8	OK 9	OK10
SiO ₂	54.69	55.28	55.11	59.07	57.46	55.39	53.98	58.35	59.04	55.69
V ₂ O ₅	0.31	0.33	0.3	0.25	0.31	0.32	0.53	0.24	0.24	0.48
Cr_2O_3	0.05	0.08	0.08	0.08	0.05	0.08	0.12	0.05	0.05	0.13
MnO	0.03	0.04	0.04	0.03	0.03	0.04	0.07	0.03	0.03	0.07
Fe ₂ O ₃	3.61	2.77	3.07	2.2	2.74	3.03	4.78	4.09	3.18	3.97
Co_3O_4	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01
Nb ₂ O ₃	0.02	0.03	0.03	0.02	0.02	0.03	0.06	0.02	0.03	0.06
P ₂ O ₅	-	-	0.03	-	0.06	-	0.17	-	0.02	0.09
SO ₃	0.16	0.44	0.33	0.54	0.47	0.17	0.68	0.34	0.3	0.22
CaO	0.05	0.13	0.06	0.07	0.11	0.02	0.16	0.07	0.09	0.1
Al ₂ O ₃	35.23	34.39	34.37	32.29	33.18	34.64	30.97	31.39	31.16	30.6
Zn0	0.01	0.01	0.01	0.01	-	0.01	0.01	0.01	-	0.01
TiO ₂	4.35	5.08	4.89	3.88	4.09	4.82	6.36	4.16	4.6	6.54
K ₂ O	0.3	0.24	0.36	0.23	0.26	0.23	0.18	0.1	0.11	0.15
Na ₂ O	0.21	0.68	0.52	0.06	0.38	0.11	0.06	0.08	0.04	0.41
LOI	2.42	3.42	3.08	2.06	3.14	2.49	3.28	2.16	2.66	2.36

Table 1 Chemical Analysis (Concentration in % Wt Oxides) for Umuoke (Ok) Clays

3.2. Mineralogical Composition

The table below shows the results of the mineralogical analysis. X-ray diffraction analyses demonstrate the Umuoke clays to be mineralogically dominated by quartz (SiO₂) and kaolinite (Al₂Si₂O₅(OH)₄), which on average constitute over 80% of the deposit (Table 2, Figure 2). Quartz levels range between 44.9-65.9%, while kaolinite varies from 18.9-37.0% across the samples. Meanwhile, minor accessory phases identified include orthoclase feldspar, illite, hematite, albite, muscovite, chlorite, and anatase. Among the accessory minerals, hematite is the most abundant, reflecting the elevated Fe₂O₃ levels noted previously. Overall, the mineralogical uniformity mirrors the observed geochemical consistency, further confirming the geological homogeneity of the Umuoke clays.

Sample location	Code	K(%)	Qz(%)	0r(%)	An(%)	Ch(%)	I(%)	Mu(%)	He(%)	Al(%)
Sample 1	0K 1	36.1	47.0	11.3	16.0	13.0	0.9	6.8		
Sample 2	OK 2	31.0	56.0	5.8		1.8	0.5	3.6	1.54	
Sample 3	OK 3	27.0	53.0	1.0		4.0		1.0		7.2
Sample 4	OK 4	31.0	57.0	1.0		4.0		1.0		5.0
Sample 5	OK 5	37.0	55.0	3.2			1.3	2.3	0.9	
Sample 6	0K 6	28.0	58.0	0.35			1.6	6.1	1.4	4.9
Sample 7	0K 7	30.6	44.9	7.8			4.15	4.69	1.97	6.0
Sample 8	OK 8	20.0	55.0	8.1			3.9	5.0	2.2	6.3
Sample 9	OK 9	37.0	50.0	-		1.5		11.1		0.4
Sample 10	OK 10	18.9	65.9	3.44			4.10	6.40	0.56	0.76

Table 2 Mineralogical Composition of the Studied Clays for Umuoke (OK)

K = Kaolinite, Or = Orthoclase An = Anatase; Ch = Chlorite Qz = Quartz; He = Hematite; I = illite Mu = Muscovite, Al = Albite



Figure 2 Clay Mineral Occurrences in the Samples in Umuoke(OK)

3.3. Plasticity

Atterberg limit tests conducted on the Umuoke samples reveal liquid limits ranging from 33.0-88.0%, plasticity limits of 10.3-47.7%, and plasticity index values between 17.4-43.5% (Table 3). Based on the Casagrande plasticity chart, the majority of Umuoke clays classify as intermediate to highly plastic inorganic clays. Exceptions include sample OK3 which plots below the A-line as a silty clay, while OK1 falls on the boundary between intermediate and low plasticity zones.

Table 3 Results of the Atterberg Limit of Umuoke (Ok) Clay

Sample code	Liquid limit	Plastic limit (PL %)	Plastic index (PI %)
OK1	33.0	10.3	28.7
0K2	44.3	25.5	18.8
0К3	60.1	42.7	17.4
0K4	69.2	25.9	43.5
0К5	47.3	28.4	18.9

0K6	42.0	21.4	20.6
0K7	45.0	13.9	31.1
0K8	88.0	47.7	40.3
0К9	49.5	31.0	18.5
OK10	39.8	11.5	28.3
Average	51.82	25.83	26.61

3.4. Physical Properties

Physical testing reveals the Umuoke clays to possess average apparent porosity, water absorption capacity, bulk density and apparent specific gravity values of 21.33%, 13.65%, 1.67 g/cm³ and 2.18, respectively (Table 4, Figure 3). Porosity ranges from 12.96-28.33% across samples, while water absorption varies between 10.16-16.38%. Higher porosity promotes greater water absorption, as exemplified by sample OK2.

Table 4 Result of Apparent Porosity, Water of Absorption, Bulk Density, and Apparent Specific Gravity of Umuoke(OK)Clay

Sample Location	Apparent porosity (%)	Water of absorption (%)	Bulk density (g/cm3)	Apparent specific gravity
0K 1	12.96	10.16	1.67	2.01
OK 2	28.33	16.38	1.73	2.41
OK 3	18.39	10.48	1.76	2.15
0K 4	24.25	14.36	1.67	2.20
OK 5	12.98	14.5	1.63	2.13
0K 6	26.24	14.99	1.75	2.57
0K 7	26.76	16.3	1.64	2.24
OK 8	19.43	12.62	1.54	1.91
OK 9	19.33	11.99	1.61	2.00
OK 10	24.64	14.75	1.67	2.22
Average	21.33	13.65	1.67	2.18



Figure 3 Chart of the Bulk Density, Water of Absorption, Apparent Porosity, Apparent Specific Gravity of Umuoke(OK) Clay

3.5. Firing Shrinkage

Analysis of drying and firing shrinkages reveals total linear shrinkage values for Umuoke clays between 6.5-15.0% (Table 5, Figure 4). On average, clays contracted by 5.16% during oven-drying at 110°C, and a further 5.74% upon firing to 1100°C. The high plasticity index of these clays translates to increased shrinkage tendencies, as expanding lattice layers between clay platelets are expelled upon water loss and dehydroxydation.

Sample Code	Dry shrinkage (%)	Fired shrinkage (%)	Total shrinkage (%)
0K 1	4.50	7.22	12.60
OK 2	5.20	2.22	7.70
ОК 3	5.80	6.58	12.00
OK 4	6.00	4.26	6.50
OK 5	5.00	5.68	10.40
0K 6	4.90	4.52	9.20
OK 7	4.00	4.67	8.00
OK 8	5.00	9.47	14.00
ОК 9	7.00	9.14	15.00
OK 10	4.20	3.65	19.20
Average	5.16	5.74	11.46

Table 5 Results of the Linear Shrinkage Tests of Umuoke(OK) Clay



Figure 4 Chart of the Linear Shrinkage of the Umuoke(OK) Clays

3.6. Cold Crushing Strength

The structural strength of Umuoke clay samples fired to 1100°C was evaluated via cold crushing tests. Strength values between 11.33-31.33 N/mm² were recorded, with an average of 20.73 N/mm2 (Table 6, Figure 5). These strength levels exceed the minimum industry standards of 15 N/mm2 specified for fireclay bricks (15). Thus, the Umuoke clays demonstrate adequate strength upon firing to be capable of serving as construction material.

Sample code	Crushing strength (N/mm ²)
OK1	24.67
OK2	16.00
0K3	20.00
OK4	20.00
OK5	23.33
OK6	16.67
OK7	18.00
OK8	26.00
0К9	31.33
OK10	11.33
Average	20.73

Table 6 Result Compressive Strength or cold Crushing of Umuoke (OK) Clays



Figure 5 Chart of the Cold Crushing Strength (N/mm2) of Umuoke (OK) Clays

3.7. Refractoriness

Refractoriness estimates derived from the bulk chemical composition of the Umuoke clays average 1680.22°C. This high refractoriness temperature signifies the Umuoke deposits comprise of minerals with elevated melting points, including quartz, kaolinite and hematite. Hence, the Umuoke clays or ceramics produced from them should withstand prolonged exposure to temperatures nearing 1700°C without undergoing softening or deformation.

4. Discussion

The mineralogical and geochemical properties of the Umuoke clay deposits in Obowo, Southeastern Nigeria were assessed to determine their suitability for various industrial applications. The samples were characterized using X-ray fluorescence (XRF) spectrometry, X-ray diffraction (XRD), and evaluation of physical properties.

4.1. Geochemical Composition

The chemical composition of the Umuoke clay samples showed a predominance of silica (SiO_2) and alumina (Al_2O_3) , which is typical for aluminosilicate clays (16–18). The silica content ranged from 53.98% to 59.07%, averaging 56.41%, while the alumina content ranged from 30.60% to 35.23%, averaging 32.82%. These high silica and alumina levels are comparable to other Nigerian clay deposits reported in studies by Abdullahi and Audu(19), Falodun et al. (20), and Oluseyi et al. (21). The relative proportion of silica to alumina $(SiO_2/Al_2O_3 ratio)$ affects the physicochemical properties and applications of clays (22). Clays with ratios between 2-4 often exhibit favorable refractory characteristics, plasticity,

and ceramic strength (23). The average SiO2/Al2O3 ratio for the Umuoke clays was 1.72, indicating moderately good refractory potential.

Apart from silica and alumina, the Umuoke clays contained several other oxides including iron, alkali metals, and alkaline earth metals. The Fe2O3 content, ranging from 2.20% to 4.78% (average 3.34%), was relatively high. While some iron oxides naturally occur in clays, excessive amounts can adversely impact the coloration and high temperature performance (24). The alkali oxides of sodium and potassium along with alkaline earths like calcium indicate the presence of some feldspathic minerals. Feldspars act as fluxes, promoting liquid phase formation and densification during firing (25). The Umuoke clays showed 0.22% K2O and 0.26% Na2O on average.

The loss on ignition (LOI) values, indicating the organic matter content, was low for the Umuoke clays ranging from 2.06% to 3.42% (average 2.71%). Minimal LOI is favorable for ceramics as high organics can cause defects during firing (26). Overall, the chemical composition highlights the aluminosilicate nature of the Umuoke clays, with high SiO₂, Al₂O₃ and moderately high Fe₂O₃ along with some fluxing oxides.

4.2. Mineralogical Composition

The mineralogical analysis by XRD showed the Umuoke clays were predominantly composed of the clay minerals kaolinite and illite along with non-clay minerals including quartz, feldspars, micas, and metal oxides. Kaolinite ranged from 18.90% to 37.0% among the samples, while illite occurred at 0.5% to 4.15%. The presence of kaolinite and illite as major minerals is typical for many clays worldwide (27,28) and concurs with studies on Nigerian clays (19,29). Kaolinite imparts favorable plasticity, workability, and strength for shaping ceramic products, while illite enhances dry mechanical strength (30,31). The Umuoke kaolinite content compares well with the range of 25-80% seen in ball clays (32).

Among the non-clay minerals, quartz was most abundant ranging from 44.9% to 65.9%. Quartz can help reduce shrinkage and warping tendencies but may also lead to cracking if cooling rates differ considerably during firing (33). Orthoclase feldspar was found from 0.35% to 11.3%, while albite occurred at up to 7.2% in some samples. Feldspars aid vitrification and liquid phase formation leading to high fired strength (25). The Umuoke clays showed higher feldspar levels compared to other Nigerian kaolinitic clays with less than 5% feldspar (29). Other non-clay minerals detected include muscovite (1.0% to 11.1%), chlorite (up to 13.0%), and the iron oxide hematite (0.56% to 2.2%). Overall, the Umuoke clay mineralogy appears well suited for manufacturing ceramic products.

4.3. Physical Properties

The physical properties of the Umuoke clays provide insight into their plasticity, working behavior, and fired characteristics. The Atterberg limits, which measure the plastic and liquid limits, showed high plasticity for the Umuoke clays based on the Casagrande plasticity chart classification. The plasticity index ranged from 25.83% to 29.03%, averaging 26.61%, while the liquid limit was 51.82% on average. High plasticity arises due to the fine particle size, platelet shape, and high surface area of the clay minerals (30). Clays with high plasticity exhibit good moldability and forming behavior, making them adequate for manufacturing bricks, tiles, ceramics, etc. which require shaping (34,35). However, high plasticity also causes increased shrinkage during drying and firing.

The Umuoke clays showed favorable porosity characteristics, which allow the escape of decomposed gases during firing thereby enabling vitrification and densification. Apparent porosity ranged from 12.96% to 28.33%, averaging 21.33%, which agrees with the standard range of 15-30% for refractory clays (36,37). Higher porosity benefits insulation refractories by providing low thermal conductivity (38).

The linear shrinkage, measuring the dimensional changes on drying and firing, lengths ranged from 6.5% to 19.2%, averaging 11.46%. This is close to the ideal 7-10% shrinkage for refractory fireclay bricks (15). Slight additions of non-plastic grog material may help tailor the shrinkage. The bulk density varied from 1.54 to 1.76 g/cm3 averaging 1.67 g/cm3, which is at the lower end of typical values of 1.7 - 2.1 g/cm3 for dense fireclay refractories (39). Lower density causes increased shrinkage.

The cold crushing strength, which indicates the load bearing capability after firing, ranged from 11.33 to 31.33 N/mm² averaging 20.73 N/mm². This meets the minimum requirement of 15 N/mm2 specified for fireclay bricks (15,40). Good strength develops from densification and liquid phase sintering during firing. Based on the chemical composition using Shuen's formula, the predicted refractoriness was 1680.22°C which agrees with the standard range of 1500–1700°C for refractory fireclays. Overall, the physical properties highlight the high plasticity of the Umuoke clays coupled with adequate porosity, shrinkage, strength, and refractoriness for utilization in ceramics and refractories.

4.4. Suitability for Industrial Applications

The chemical, mineralogical, and physical properties were assessed to determine the suitability of the Umuoke clays for various industrial applications, especially in the ceramics and refractories sector. Table 7 below summarizes the key characteristics based on the analytical results.

Property	Observed	Industrial relevance
Chemical		
SiO2	53.98 - 59.07%	Good refractory oxide
Al203	30.60 - 35.23%	Imparts refractory nature
Fe2O3	2.20 - 4.78%	High iron reduces refractory quality
Flux oxides	Present	Improves vitrification
LOI	2.06 - 3.42%	Low organics, good for ceramics
Mineralogy		
Kaolinite	18.90 - 37.0%	Imparts plasticity and strength
Illite	0.5 - 4.15%	Enhances dry mechanical strength
Quartz	44.9 - 65.9%	Reduces shrinkage but can cause cracking
Feldspar	0.35 - 16%	Promotes vitrification and liquid phase
Physical properties		
Plasticity	High	Good moldability and shaping
Porosity	12.96 - 28.33%	Aids refractory insultation
Shrinkage	6.5 - 19.2%	Mostly in ideal range for refractories
Bulk density	1.54 – 1.76 g/cm3	Lower density causes high shrinkage
Strength	11.33 - 31.33 N/mm2	Meets minimum standard for fireclay bricks
Refractoriness	1680.22°C	Suitable for high temperature applications
Potential applications		
- Refractory bricks	- Tableware ceramics	
- Insulation firebricks	- Architectural ceramics	
- Ceramic tiles	- Sculpture clay bodies	

Table 7 Summary of Umuoke clay properties and potential applications

The Umuoke clays demonstrate various properties favorable for refractory materials used in high temperature applications, such as refractory bricks, insulation firebricks, and refractory linings (38,41). The high SiO₂ and Al₂O₃ content coupled with the moderate refractoriness of 1680°C indicates suitability to withstand elevated temperatures without deformation. The porosity meets standards for insulation refractories. The strength satisfies minimum requirements for fireclay bricks which are exposed to abrasion and load. However, the high Fe_2O_3 content needs to be mitigated via mineral processing to avoid coloration issues and further improve the refractory quality.

Additionally, the Umuoke clays also possess characteristics appropriate for structural and traditional ceramics which require shaping and firing at relatively lower temperatures than refractories. The fine grain size and high plasticity arising from the kaolinite and illite minerals enables easy molding into complex shapes. The fluxing oxides help vitrification while quartz acts as a filler reducing shrinkage. After firing, the ceramics develop adequate strength from sintering and liquid phase formation. Hence, the Umuoke clays can be suitable for manufacturing ceramic tableware, architectural bricks, wall/floor tiles, pottery, statues, and artware (42). While the working characteristics are favorable,

the high shrinkage needs control through processing adjustments. Blending with non-plastic materials like grog or chamotte can help curtail shrinkage.

5. Conclusions

This study characterized the mineralogical and geochemical properties of clay deposits from Umuoke, Obowo in southeastern Nigeria. The aim was to evaluate their suitability as local raw materials for various industrial applications, especially in the ceramics and refractories sector. Ten clay samples collected from different mining pits across Umuoke were analyzed using X-ray fluorescence (XRF), X-ray diffraction (XRD), and laboratory tests of key physical properties.

The findings from this study demonstrate the suitability of Umuoke clay deposits in Obowo, Southeastern Nigeria for applications in the ceramics and refractories industry, with potential utilization in refractory bricks, ceramic tableware, architectural ceramics, wall tiles, pottery items etc. The chemical composition showed predominantly aluminosilicate nature with high SiO₂, Al₂O₃ and moderately high Fe₂O₃. Clay minerals kaolinite and illite occur along with non-clay minerals including quartz, feldspars, mica and metal oxides. The clays exhibit high plasticity coupled with appropriate fired porosity, strength, and refractoriness. While the properties are mostly favorable, minor mineral processing and blending steps can help mitigate issues like high iron content, shrinkage, and cracking during firing. The local availability of the Umuoke clays can promote indigenous ceramics manufacturing, substituting imports and fueling small-scale enterprises. Further pilot-scale testing will help establish optimal process parameters and applications.

However, certain limitations need mitigation. The high Fe_2O_3 content can lead to undesirably dark colors in ceramics. Hence mineral processing is required to lower the iron levels. Blending with non-plastic diluents such as feldspar and grog may help control the high shrinkage observed. Cracking and warping could also occur during drying and firing due to quartz inversion combined with elevated shrinkage. Introducing appropriate grog additions can minimize these defects by managing shrinkage rates.

Further research can build on these foundational results. Pilot-scale testing should establish optimal formulations and process parameters for targeted applications through empirical adjustments. Additionally, comprehensive nationwide deposit prospecting is recommended to quantify total reserves and spatial distribution thereby facilitating organized exploitation.

Compliance with ethical standards

Acknowledgement

- National Steel and Raw Materials Exploration Agency, Kaduna. Nigeria for carrying out the geochemical and mineralogical analyses
- Soil Laboratory and Concrete Laboratory, Civil Engineering Department, Akanu Ibiam Federal Polytechnic Unwana, Nigeria. for carrying out the Atterberg limit tests and Cold crushing tests in work
- Ceramic and Glass Technology Department, Akanu Ibiam Federal Polytechnic Unwana, Nigeria. for carrying out some of the physical tests in thi work

Disclosure of conflict of interest

The author, Thaddeus C. Azubuike, declares no conflicts of interest.

References

- [1] Wilson MJ. The origin and formation of clay minerals in soils: past, present and future perspectives. Clay Miner. 1999 Mar;34(1):7–25.
- [2] Afolabi RO, Orodu OD, Efeovbokhan VE. Properties and application of Nigerian bentonite clay deposits for drilling mud formulation: Recent advances and future prospects. Appl Clay Sci. 2017 Jul;143:39–49.
- [3] Abuh M, Abia-Bassey N, Udeinya T, Nwannewuihe A, Agwul A, Akpomie K. Industrial Potentials of Adiabo clay in Calabar municipal of Cross River State South South Nigeria. Pac J Sci Technol. 2014 Jan 1;15:63–75.
- [4] Bukalo NN, Ekosse GIE, Odiyo JO, Ogola JS. Geochemistry of Selected Kaolins from Cameroon and Nigeria. Open Geosci [Internet]. 2017 Dec 2 [cited 2023 Aug 31];9(1). Available from: https://www.degruyter.com/document/doi/10.1515/geo-2017-0045/html

- [5] Irabor PS. Physical and Chemical Investigation on Some Nigerian Kaolinite Clays for Use in the Ceramics and Allied Industries. Niger J Eng Res Dev. 2002;1(1):54–9.
- [6] Lawal O, Abdullahi Y. Evaluation of industrial potentials of alluvial clays from the confluence of rivers Niger and Mimi. Sci World J. 2010 Jan 1;5(3):17–20.
- [7] Perkins D. Mineralogy. 2nd edition. New Delhi: PHI Learning Private Ltd; 1998. 483 p.
- [8] Khurana IS, Kaur S, Kaur H, Khurana RK. Multifaceted role of clay minerals in pharmaceuticals. Future Sci OA. 2015 Nov;1(3):fso.15.6.
- [9] Galindo AL, Viseras C, Cerezo P. Compositional, technical and safety specifications of clays to be used as pharmaceutical and cosmetic products. Appl Clay Sci. 2007 Apr;36(1–3):51–63.
- [10] Anumata SE, Ogwuegbu MOC, Nwoko CIA, Akalezi CO. Mineralogical and Elemental Analysis of Clay Sample from Iyinwaogba Umuosode Alike Ikenanzizi Obowo Local Government Area of Imo State. J Mater Environ Sci. 2021;12(4):511–31.
- [11] Nwajide CS. Geology of Nigeria's Sedimentary Basins. Lagos: CSS Bookshops Limited Brood Street; 2013.
- [12] Ogala JE. The geochemistry of lignite from the Neogene Ogwashi-Asaba Formation, Niger Delta Basin, southern Nigeria. Earth Sci Res J. 2012 Dec;16(2):151–64.
- [13] Emofuriefa W, Kayode A, Coker S. Mineralogy, Geochemistry and Economic Evaluation of the Kaolin Deposits Ubulu-Uku Awo-Omama and Buan in Southern Nigeria. J Min Geol. 1992;28:211–20.
- [14] Nton M, Elueze AA. Composition Characteristics and Industrial Assessment of Sedimentary Clay bodies in part of Eastern Dahomey basin, Southwestern Nigeria. J Min Geol. 2005 Sep 1;4:175–84.
- [15] Yakubu S, Ibrahim MY. Suitability of Birnin Gwari and Maraban Rido Clays as Refractory Materials. Am J Eng Res. 2014 Jan 1;03:11–25.
- [16] Amar A, Kali A, Loulidi I, Cho Y, Sadoq M, Hadey C, et al. Unveiling the potency of Silica-Alumina-rich clay in phenol remediation and its repurposing prospects. Inorg Chem Commun. 2023 Aug 1;154:110983.
- [17] Chesworth W, Camps Arbestain M, Macías F, Spaargaren O, Spaargaren O, Mualem Y, et al. Clay Minerals: Silicates. In: Chesworth W, editor. Encyclopedia of Soil Science [Internet]. Dordrecht: Springer Netherlands; 2008 [cited 2023 Aug 16]. p. 141–4. (Encyclopedia of Earth Sciences Series). Available from: https://doi.org/10.1007/978-1-4020-3995-9_110
- [18] Matsui MS, Carle T, Costa M. Chapter 18 The interaction of metals and the skin: The good, bad, and the ugly. In: Nordberg GF, Costa M, editors. Handbook on the Toxicology of Metals (Fifth Edition) [Internet]. Academic Press; 2022 [cited 2023 Aug 16]. p. 407–20. Available from: https://www.sciencedirect.com/science/article/pii/B9780128232927000115
- [19] Abdullahi SL, Audu AA. Comparative analysis on chemical composition of bentonite clays obtained from Ashaka and tango deposits in Gombe State, Nigeria. ChemSearch J. 2017;8(2):35–40.
- [20] Falodun O, Folorunso D, Oke S, Borode JO. Investigating the Effects of High Alumina Cement and Silica Sand on the Suitability of Ikere Ekiti Clay for Refractory Applications. Am J Eng Res. 2017 Jul 1;6:24–8.
- [21] Oluseyi A, Aladesuyi O, Pal M, Das S. Evaluation of Nigerian Source of Kaolin as A Raw Material for Mullite Synthesis. Orient J Chem. 2016 Jun 28;32:1571–82.
- [22] Partyka J, Sitarz M, Leśniak M, Gasek K, Jeleń P. The effect of SiO2/Al2O3 ratio on the structure and microstructure of the glazes from SiO2–Al2O3–CaO–MgO–Na2O–K2O system. Spectrochim Acta A Mol Biomol Spectrosc. 2015 Jan 5;134:621–30.
- [23] Chesti AR. Refractories manufacture, properties and applications. New Delhi: Prentice Hall of India; 1986. 180 p.
- [24] Ekosse GIE. Kaolin deposits and occurrences in Africa: Geology, mineralogy and utilization. Appl Clay Sci. 2010 Oct 1;50(2):212–36.
- [25] Breuer S. RSC Education. 2012 [cited 2023 Aug 19]. The chemistry of pottery. Available from: https://edu.rsc.org/feature/the-chemistry-of-pottery/2020245.article
- [26] Hoogsteen MJJ, Lantinga EA, Bakker EJ, Tittonell PA. An Evaluation of the Loss-on-Ignition Method for Determining the Soil Organic Matter Content of Calcareous Soils. Commun Soil Sci Plant Anal [Internet]. 2018; Available from: https://doi.org/10.1080/00103624.2018.1474475s

- [27] Dill HG. A geological and mineralogical review of clay mineral deposits and phyllosilicate ore guides in Central Europe A function of geodynamics and climate change. Ore Geol Rev. 2020 Apr 1;119:103304.
- [28] Rat E, Martínez-Martínez S, Sánchez-Garrido JA, Pérez-Villarejo L, Garzón E, Sánchez-Soto PJ. Characterization, thermal and ceramic properties of clays from Alhabia (Almería, Spain). Ceram Int. 2023 May 1;49(9, Part B):14814–25.
- [29] Elueze AA, Ntekim EE, Ekwere SJ. Compositional and Industrial Assessment of Clay bodies in the area, south western Nigeria. J Min Geol. 1999;33(2):117–24.
- [30] Andrade FA, Al-Qureshi HA, Hotza D. Measuring the plasticity of clays: A review. Appl Clay Sci. 2011 Jan 1;51(1):1–7.
- [31] Supandi S, Zakaria Z, Sukiyah E, Sudradjat A. The Influence of Kaolinite Illite toward mechanical properties of Claystone. Open Geosci. 2019 Aug 29;11:440–6.
- [32] Galos K. Composition and ceramic properties of ball clays for porcelain stoneware tiles manufacture in Poland. Appl Clay Sci. 2011 Jan 1;51(1):74–85.
- [33] Earle S, Earle S. 2.4 Silicate Minerals. 2015 Sep 1 [cited 2023 Aug 19]; Available from: https://opentextbc.ca/geology/chapter/2-4-silicate-minerals/
- [34] Aubeny C, Lytton R. Properties of High-Plasticity Clays [Internet]. Texas; 2002 Jan. Report No.: 2100–2. Available from: https://static.tti.tamu.edu/tti.tamu.edu/documents/2100-2.pdf
- [35] Hossain S, Islam A, Badhon FF, Imtiaz T. Atterberg Limit Test. In: Properties and Behavior of Soil Online Lab Manual [Internet]. Mavs Open Press; 2021 [cited 2023 Aug 21]. Available from: https://uta.pressbooks.pub/soilmechanics/chapter/atterberg-limit-test/
- [36] Al-Amaireh M. Production of FireClay Refractory Bricks from Local Materials. 2009 Jan 1;26.
- [37] Musa U, Aliyu A, Sadiq MM, Mohammed IA, Isah A, Mohammed AD. A Comparative Study of the Refractory Properties of Selected Clays in Niger State, Nigeria. Adv Biotechnol Chem Process - World Open Access J. 2012 Dec 14;2:1–5.
- [38] Chesters JH. Refractories: production and properties. London: The Iron and Steel Institute; 1973. 553 p.
- [39] Yami A, Hassan M, Umaru S. Evaluation of The Refractory Characteristics of Dukku Clay Deposit. Cont J Eng Sci. 2007 Jan 1;Vol.2.:Pp. 15-21.
- [40] ASTM. Annual book of ASTM standards, 1982. Philadelphia: American Society for Testing and Materials; 1982.
 24 p.
- [41] Gilchrist JD. Fuels, furnaces, and refractories [Internet]. New York: Pergamon Press; 1977 [cited 2023 Aug 25].
 353 p. Available from: http://archive.org/details/fuelsfurnacesref0000gilc
- [42] Grimshaw RW. The Chemistry and physics of clays and allied ceramic materials. 4th ed., rev. London: Benn; 1971.